

## PHYS1221-1231 tutorials

### Tutorial 10 Answers

$$Q1 \quad \Delta E \cdot \Delta t = h/2\pi. \quad E = -\left(\frac{me^4}{8\epsilon_0^2 h^2}\right) \frac{1}{n^2}$$

$$\Delta E/E = \frac{h}{2\pi \Delta t \cdot E} = 20 \text{ parts per billion. } 8 \text{ significant figures only}$$

$$Q2 \quad \Delta E \cdot \Delta t = h/2\pi. \text{ So } \Delta E/E = \frac{h}{2\pi \Delta t \cdot E} = 2 \text{ parts in } 10^{10}.$$

$$Q3 \quad \text{The photons must have } \lambda \sim 10 \text{ pm, so they have } E = hc/\lambda = 120 \text{ keV} \text{ and } p = h/\lambda = 7 \cdot 10^{-23} \text{ kgm.s}^{-1}$$

The binding energy of the hydrogen atom is only 13 eV so interactions with these photons will ionise it.

The momentum of Sommerfeld electrons is  $\frac{m}{2\epsilon_0} \frac{e^2}{nh} = 2 \cdot 10^{-25} \text{ kgm.s}^{-1} \ll p_{\text{photon}}$  so the collisions would give the electrons large recoil velocities.

$$Q4 \quad \Delta f \cdot \Delta t \geq 1/2\pi. \quad \Delta f \geq 0.16 \text{ Hz. For the second case, } 1.6 \text{ Hz.}$$

Answers for musicians: At this pitch, the uncertainty is half a semitone. (In practice, this is only 5 vibrations of the string, and it takes at least a few cycles for the bow to set up periodic Helmholtz motion, so it won't have a clear pitch at all. Musicians know this: in very fast runs in the bass you can get away with a lot: the ear will interpolate.)

Serves the composer right for writing low demisemiquavers for the basses – but then Schubert never finished that symphony: Detective we're re-opening the case – I want you to interview all the bassists who were in Vienna on or around the night of 18 November 1828.)

$$Q5 \quad \text{For electrons, } I d\theta \text{ would be proportional to the number of electrons detected between the angles } \theta \text{ and } \theta+d\theta. \text{ The integral of } I \text{ from } \theta = -\pi/2 \text{ to } \pi/2 \text{ would be set equal to } N, \text{ and this equation would define } I_0, \text{ the maximum in the electron distribution function.}$$

If you covered one slit, only half as many electrons would pass through. Further, there would be no interference, and only diffraction, so

$$I_{\text{single}}(\theta) = \frac{I_0}{2} \cos^2\left(\frac{\sin \alpha}{\alpha}\right)^2$$

$$Q6 \quad n_e = \frac{2}{3} \frac{8\sqrt{2} \pi m_e^{3/2}}{h^3} E_F^{3/2} = 8.5 \cdot 10^{28} \text{ electrons/m}^3.$$

atom density = density/(atomic mass) =  $\frac{(8920 \text{ kg.m}^{-3})}{(63.5 \text{ kg/kmol} / 6 \cdot 10^{26} \text{ atoms/kmol})} = 8.4 \cdot 10^{28} \text{ atoms/m}^3$ . So one conduction electron per atom.

$$v = \sqrt{\frac{2K}{m}} \cong \sqrt{\frac{2E}{m}} = \sqrt{\frac{2 \cdot 7.05 \text{ eV}}{9.11 \cdot 10^{-31} \text{ kg}}} = 1.6 \cdot 10^3 \text{ km/s} = 0.5\% \text{ of } c.$$

For a conductor,  $i = nAvq$  so  $v = i/nAq = 0.95 \text{ mm/s}$ .

If all moved at 0.5% of  $c$ ,  $i = nAvq \rightarrow 17 \text{ GA}$