

THE UNIVERSITY OF NEW SOUTH WALES  
SCHOOL OF PHYSICS

PHYS3050 NUCLEAR PHYSICS

EXAMINATION – SEPTEMBER, 2006

PAPER 1 – MID-SESSION

Time allowed = 50 minutes

Total number of questions = 4

Total number of marks = 40

Answer ALL questions

The questions are of equal value

Portable battery powered electronic calculators (without alphabetic keyboards) may be used.

The paper may be retained by the candidate

General Data:

1 unified mass unit (u) =  $931.5 \text{ MeV}/c^2$

Planck's constant  $h = 6.63 \times 10^{-34} \text{ Js}$

Boltzmann's constant  $k = 1.38 \times 10^{-23} \text{ JK}^{-1}$

Avogadro's number =  $6.022 \times 10^{23} \text{ (g-mole)}^{-1}$

Permittivity constant  $\epsilon_0 = 8.85 \times 10^{-12} \text{ Fm}^{-1}$

Fundamental charge unit  $e = 1.60 \times 10^{-19} \text{ C}$

speed of light (vacuum)  $c = 3.0 \times 10^8 \text{ m/s}$

electron mass =  $9.11 \times 10^{-31} \text{ kg} = 5.4858 \times 10^{-4} \text{ u} = 0.511 \text{ MeV}/c^2$

neutron mass =  $1.6749 \times 10^{-27} \text{ kg} = 1.008665 \text{ u} = 939.573 \text{ MeV}/c^2$

proton mass =  $1.6726 \times 10^{-27} \text{ kg} = 1.0072765 \text{ u} = 938.280 \text{ MeV}/c^2$

1 year =  $3.156 \times 10^7 \text{ s}$

### Question 1 [10 marks]

Radionuclides are useful sources of small amounts of energy in space vehicles, remote communication stations, heart pacemakers etc. Calculate the power available in Watts from a gram of  $^{210}\text{Po}$ , an  $\alpha$ -emitter with an energy of 5.30 MeV and a half-life of 138 days. [Atomic mass of  $^{210}\text{Po} = 209.982848 \text{ u}$ ].

### Question 2 [10 marks]

The lowest energy levels in the Shell Model, in order of increasing energy, are

$$1 s_{1/2}, 1 p_{3/2}, 1 p_{1/2}, 1 d_{5/2}, 2 s_{1/2}, 1 d_{3/2}, 1 f_j \dots$$

- What are the possible values of  $j$  for the  $1f$  levels.
- What is the value of  $j$  for the lowest  $1f$  level? Justify your answer!
- Determine the spin and parity of the ground states of both the  $^{40}\text{Ca}$  and  $^{41}\text{Ca}$  nuclides.
- In the Shell model, a 'spin-orbit' interaction splits all energy levels except the 's-type' levels. Why do the s-type levels remain unsplit?

### Question 3 [10 marks]

In stars slightly more massive than the Sun, hydrogen burning is carried out mainly by the CNO cycle, whose first step is  $p + {}^{12}_6\text{C} \rightarrow {}^{13}_7\text{N} + \gamma$ . Estimate the energy of the gamma (in MeV), assuming the two initial nuclei are essentially at rest. Justify any simplifying assumptions you make. [Atomic masses:  ${}^1_1\text{H} = 1.007825$ ,  ${}^{12}_6\text{C} = 12.000000$ ,  ${}^{13}_7\text{N} = 13.005739 \text{ u}$ ].

### Question 4 [10 marks]

- Describe **briefly** the 'origin' of the various terms in the Semi-Empirical Mass Formula. [NB: detailed mathematical expressions and values of constants are not required].
- Suggest a simple reason why the  ${}^{12}_6\text{C}$  nuclide has a higher binding energy (i.e. more stable) than  ${}^{12}_7\text{N}$ , even though they are isobars (i.e. same number of nucleons, A)?

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①

$$1/ \quad {}_{88}^{210}\text{Po} \quad t_{1/2} = 138 \text{ d}$$

$$E_{\alpha} = 5.3 \text{ MeV}$$

$$\begin{aligned} \text{no' nuclei} &= \frac{A_{\text{Avogadro}}}{210} \\ &= 2.87 \times 10^{21} \end{aligned}$$

$$\lambda = \frac{\ln 2}{t_{1/2}} = \frac{0.693}{138 \times 24 \times 3600} = 5.81 \times 10^{-8} \text{ s}^{-1}$$

$$\text{Activity} = \lambda N = 1.67 \times 10^{14} \text{ Bq}$$

$$\begin{aligned} \text{Power} &= \frac{\text{Energy}}{\text{second}} = 1.67 \times 10^{14} \times 5.3 \times 10^{-6} \times 1.6 \times 10^{-19} \\ &= 141 \text{ W} \end{aligned}$$

$$2/ \quad 1f_j \quad a) \quad l=3 \text{ ('f' or 'f')}$$

$$\therefore j = l \pm \frac{1}{2} = \frac{5}{2} \text{ or } \frac{7}{2}$$

b/  $1f_{5/2}$  and  $1f_{7/2}$  have different <sup>②</sup> energies due to spin-orbit effect  
 $\propto \underline{l \cdot s}$

You don't know the sign of the constant of proportionality, a priori, but you can get it from the given level scheme

$1p_{3/2}$  is lower energy than  $1p_{1/2}$

$1d_{5/2}$  is lower than  $1d_{3/2}$

$\therefore 1f_{7/2}$  is lower than  $1f_{5/2}$

c/  ${}_{20}^{40}\text{Ca}$  - even  $Z$ , even  $N$   
- all nucleons are paired

$$\therefore I^{\pi} = 0^{+}$$

$Z=20$ ,  ${}^{41}_{20}\text{Ca}$ ,  $I^\pi$  determined by the unpaired  $n$  (#21) ③

Fill the levels ( $n$ )  $2j+1$ -degenerate

$$1s_{1/2} = 2$$

$$1p_{3/2} = 4$$

$$1p_{1/2} = 2$$

$$1d_{5/2} = 6$$

$$2s_{1/2} = 2$$

$$1d_{3/2} = 4$$

20  $\therefore$  21<sup>st</sup>  $n$  is

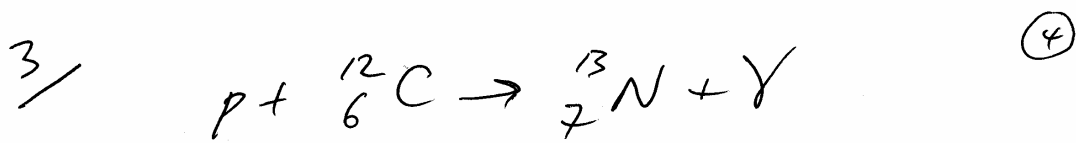
in  $1f_{7/2}$

$\therefore$  spin  $I = 7/2$

$$\text{parity} = (-1)^l = (-1)^3 = -$$

$\therefore$   ${}^{41}_{20}\text{Ca}$  is  $\frac{7}{2}^-$

$d/$  's' levels have  $l=0$  i.e. no orbital a.m.  $\therefore$  no spin-orbit effect.



can use atomic masses since no  $e^-$  cancels

$$\begin{aligned}
 Q &= (M_i - M_f) c^2 \\
 &= (1.007825 + 12.000000 - 13.005739) \text{u} \cdot c^2 \\
 &= 0.002086 \text{ u} \cdot c^2 \\
 &= 0.002086 \times 931.5 = 1.943 \text{ MeV}
 \end{aligned}$$


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4/ a. SEMF terms.

$$\begin{aligned}
 \text{Volume} &\propto A \\
 \text{Surface} &\propto -A^{2/3} \\
 \text{Coulomb} &\propto -\frac{Z^2}{A^{1/3}} \\
 \text{Symmetry} & \\
 \text{Pairing} &
 \end{aligned}$$

b.  ${}_{7}^{12}\text{N}$  has  $Z_p$  of  ${}_{6}^{12}\text{C}$  which has 6.  $\therefore$  greater Coulomb repulsion in  ${}_{7}^{12}\text{N}$ .